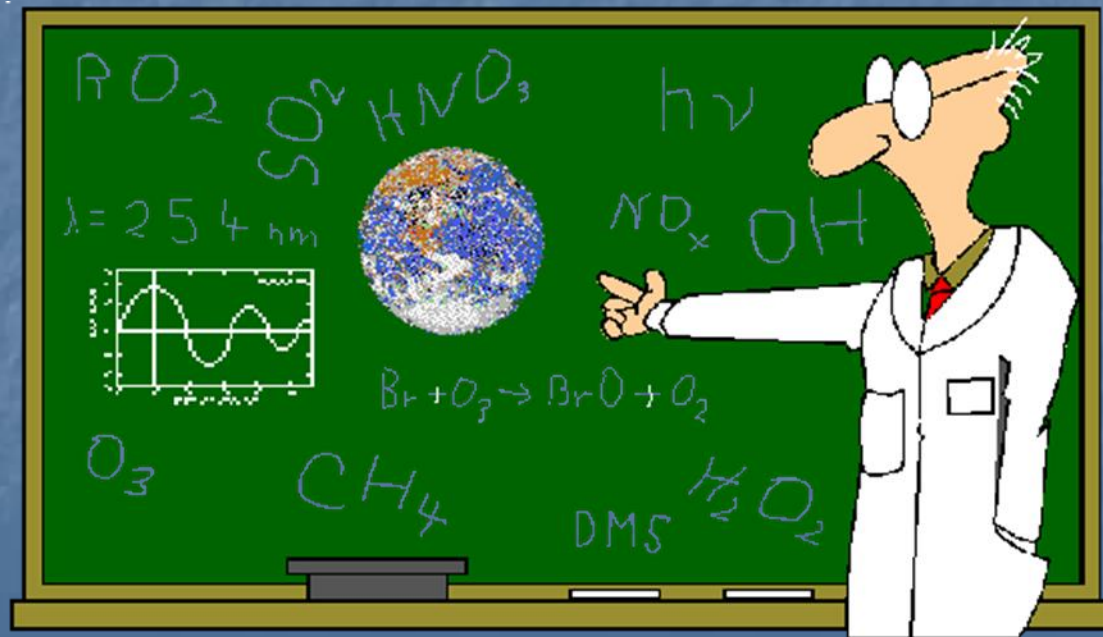


Substances of Life: Macromolecules



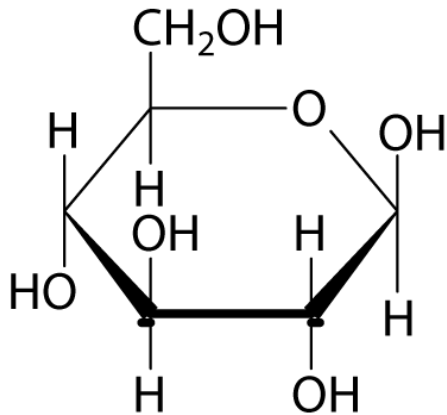


Function of the four Macromolecules of Life:

- Carbohydrates
- Lipids
- Proteins
- Nucleic Acids

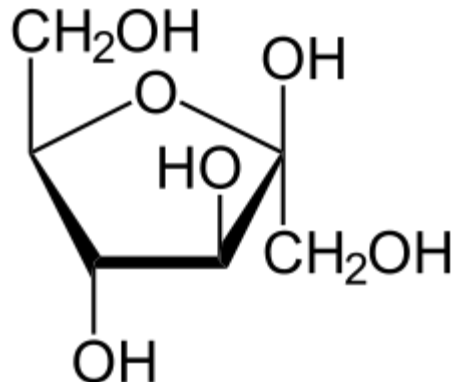
Carbohydrates

- Glucose (a monosaccharide)



**RING SHAPE,
C, H, O (1:2:1)**

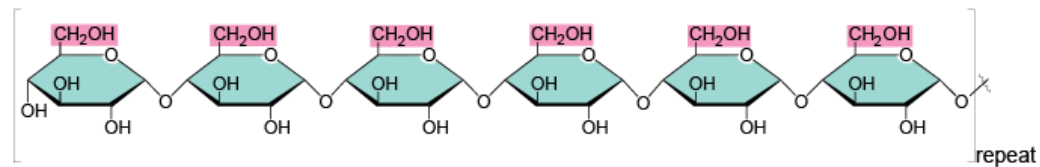
- Fructose (a monosaccharide)



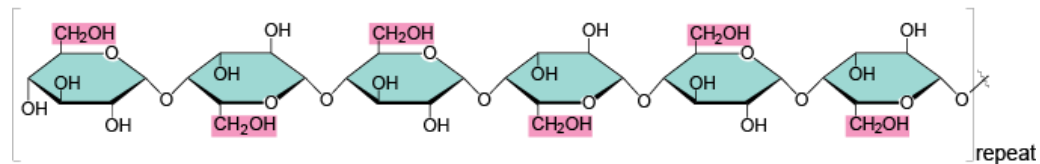
How can you recognize a carbohydrate?

- A polysaccharide is a chain of monosaccharides (more than two)!

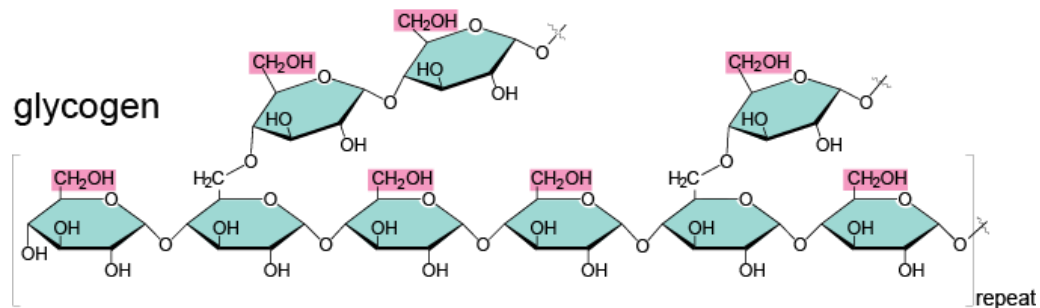
starch



cellulose



glycogen



Function of Polysaccharide Carbohydrates:

- Carbohydrates are for **QUICK energy!**
 - **Starch** – food storage in plants
 - Stores **ATP**
 - **Glycogen** – food storage in animals
 - Stores **ATP**
- They are also for **structure**
 - **Cellulose** – in cell walls of plants for support

Lipids

- Contain **C, H, O** (ratio: 1:2:low #)
- Insoluble in water (lipids are hydrophobic)
- Lipids are used for **long term energy storage, insulation, and protective coatings** (cuticle on plant leaves, blubber)
- **Contain Glycerol and Fatty Acids**

Two kinds:

Saturated and Unsaturated



Saturated Fats

- Saturated fats are saturated (surrounded) by Hydrogen, these are the fats that are considered **NOT good for you** (but you do need some to remain healthy)
- **Solid** at room temperature
- Examples are **oils, butter, animal fat, cheese, cream**

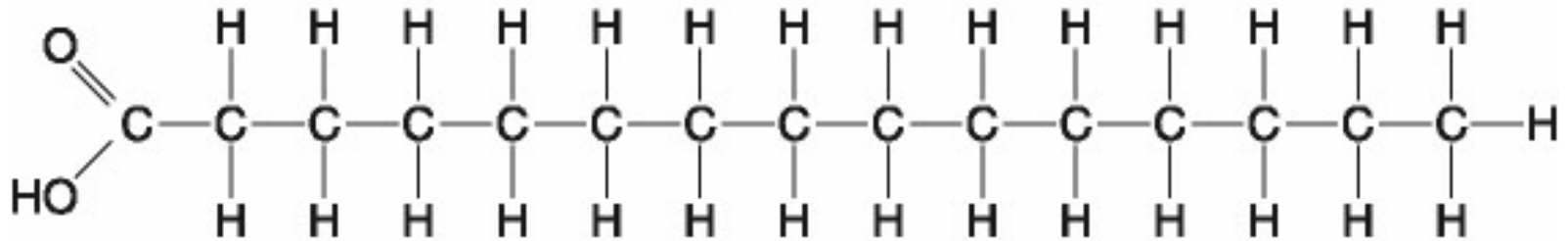


Unsaturated Fats

- Unsaturated fats are not surrounded by hydrogen, they contain double bonds and kinks in their structure. These are considered **good for you**.
- **Liquid** at room temperature
- Examples are **oils from plants** (sunflower, corn, olive), nuts, peanut butter, avocado

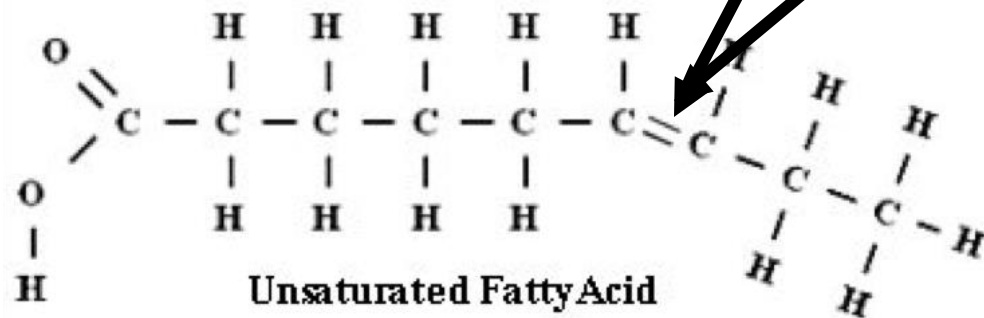
How can you recognize lipids?

- This is a **saturated** fat (think: surrounded by hydrogens, single bonds)

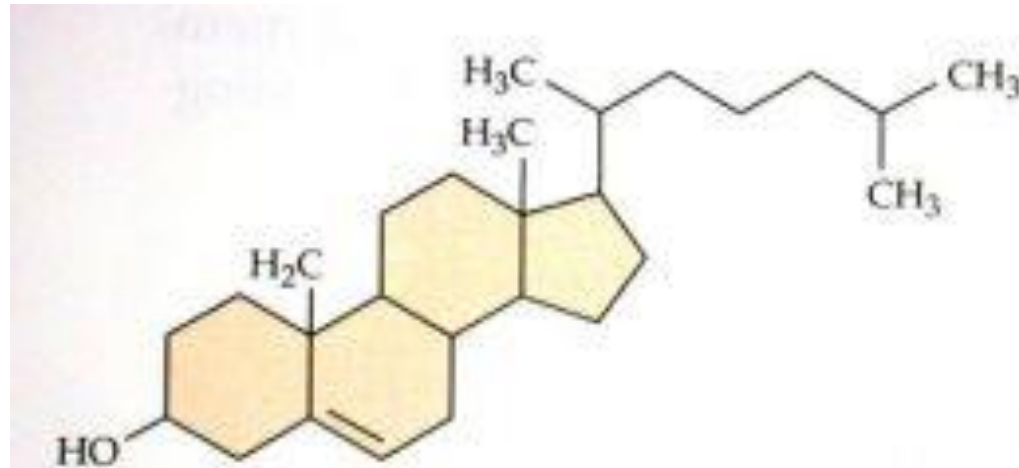


How can you recognize lipids?

- This is **unsaturated** fat (think: NOT surrounded by hydrogens, Double Bonds and sometimes kinks)



Another type of lipid - cholesterol



Cholesterol is a constituent of membranes and the source of steroid hormones.

Proteins

- Contain C, H, O, N & sometimes S
- Made up of a chain of amino acids – the order of the amino acids is important to make different proteins
- Also called polypeptides

Proteins are . . .


- The major **building blocks** of our **cells and organs**
- Used in **muscle contraction**
- Used for **transporting substances** like oxygen in the bloodstream
- Providers of **immunity**
- Used to carry out chemical reactions as **enzymes**
- **Types:** motor, structural, transport, enzymes, receptor, contractile, defensive

Enzymes

- * Most enzymes are **proteins**
- * Enzymes speed up chemical reactions in the body
 - the enzyme bonds to a **substrate** at an **active site**
 - enzyme names typically end in **-ase**
 - enzymes cause the reactions breaking down molecules or building molecules to occur at an **extremely fast rate** and **use less energy**

Nucleic Acids

- Contain C, H, O, N, P
- Made up of nucleotides (3 part units) – the order of the nucleotides is important to make different genes
- Stores genetic information in cells (DNA, RNA)



Biomolecule	Examples	Monomer form	Polymer form	How do I recognize it? Draw the molecule	Functions
Carbohydrates					
	Fats, oils, waxes				
		Amino acids			
			Deoxyribonucleic acid		