# **Biochemistry - STRUCTURE**

Biomolecules
Macromolecules
Or...major organic compounds

# Carbohydrates

are organic compounds containing carbon, hydrogen, and oxygen in the ratio 1:2:1 (carbon:hydrogen:oxygen)

are made by autotrophs

are the body's primary source of energy - QUICK ENERGY!

come in two basic forms: monomers and polymers

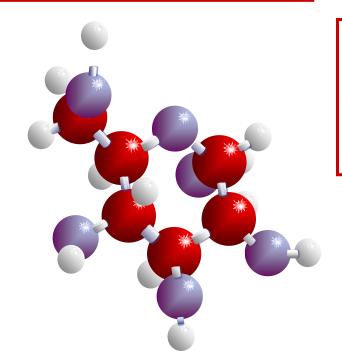
are made of monomers (building blocks) called monosaccharides



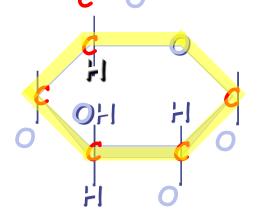
# Monosaccharides are the building blocks of carbohydrates

Glucose is a monosaccharide.  $C_6H_{12}O_6$ 

Other monosaccharides: fructose (fruit sugar) galactose (milk sugar).

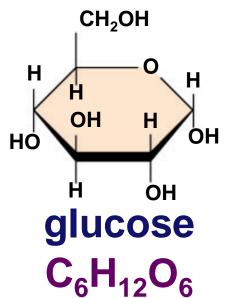


Carbohydrates will usually be in a "ring shape"



## Names for sugars usually end in <u>-ose</u>

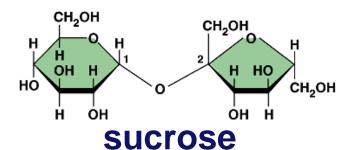
- glucose
- ◆ fructose
- **◆ sucrose**
- maltose





fructose



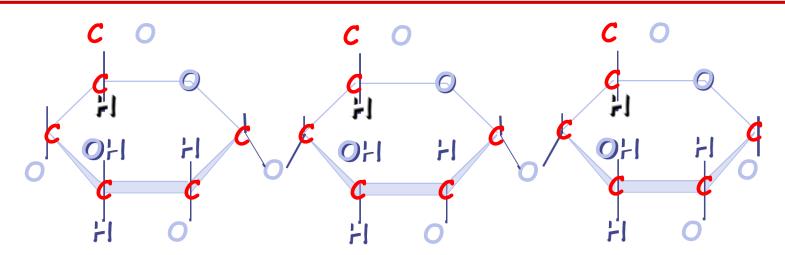


maltose



# Polysaccharides are complex carbohydrates made of long chains of monosaccharides.

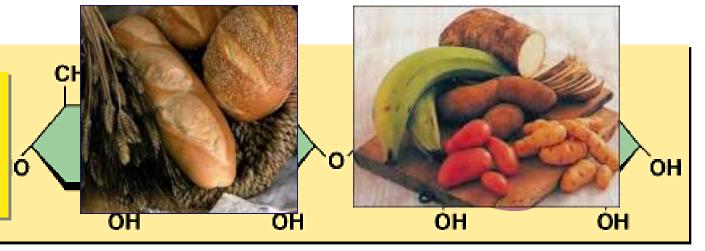
Starches (bread, cereals, and pastas) and cellulose (plant cell walls) are common sources of complex carbs.



Sucrose (table sugar) is an example of a sugar with only two monosaccharides.

# Digesting starch vs. cellulose

Starches are easy to digest



cellulose hard to digest



# Lipids

- Examples
  - ◆ fats
  - ♦ oils
  - ◆ waxes







# Lipids

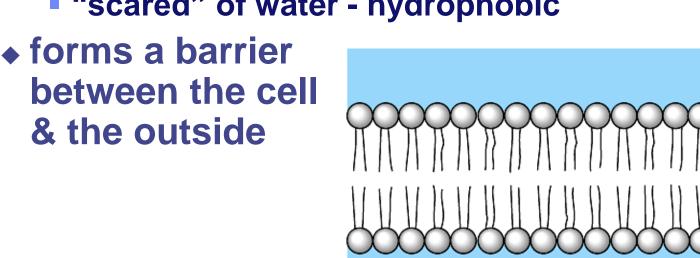
## made up of

glycerol and fatty acids

Note the molecular structure's "E" shape

# Other lipids in biology

- Cell membranes are made out of lipids
  - phospholipids
  - heads are on the outside touching water
    - "like" water hydrophilic
  - tails are on inside away from water
    - "scared" of water hydrophobic
  - & the outside





# Different types of fats

#### Diagrammatic representation of fats

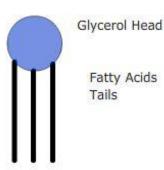
#### **Fatty Acids**

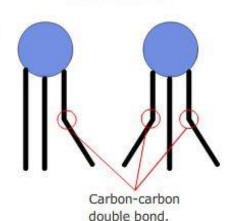
#### Saturated Fatty Acids

#### Unsaturated Fatty Acids

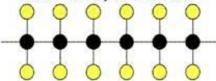


**Unsaturated Fats** 

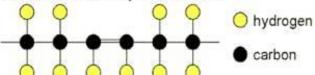




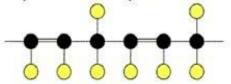
#### Saturated fatty acid molecule



#### Monounsaturated fatty acid molecule



#### Polyunsaturated fatty acid molecule



### **Proteins**

■ Monomer = <u>amino acids</u>

Over 20 different amino acids

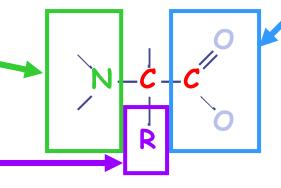
# Proteins

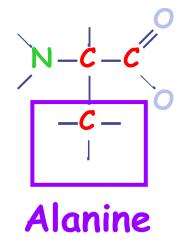
are made up of

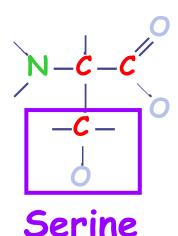
an amino group

a carboxyl group

and an "R" group which varies in the different amino acids







# Putting PROTEINS together

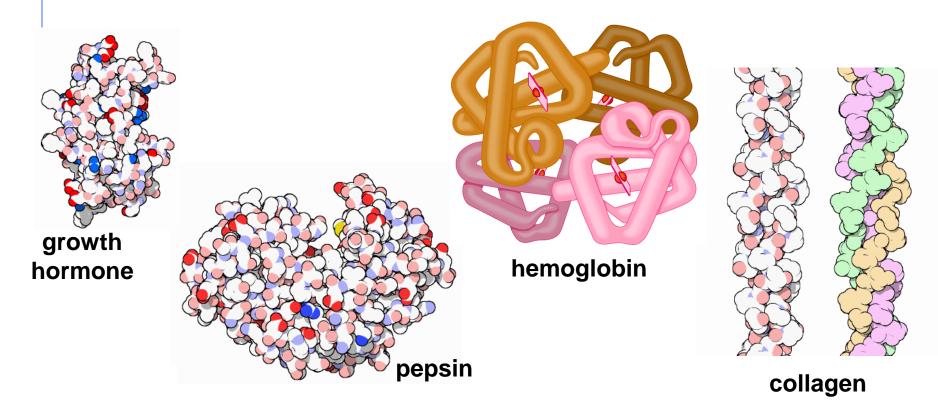
- Held together by peptide bonds
- Peptide bonds are formed through dehydration synthesis (loss of a water molecule)
- Peptide bonds can be broken by hydrolysis (addition of a water molecule

$$R \xrightarrow{O} H_2N^{-}R' \xrightarrow{R'} R \xrightarrow{N} R' + H_2O$$

Dehydration synthesis

# For proteins: SHAPE matters!

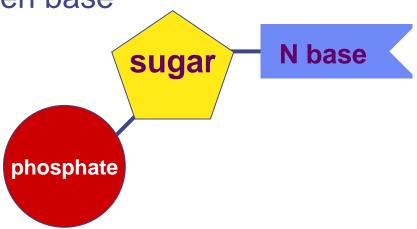
- Proteins fold & twist into 3-D shape
  - that's what happens in the cell!
- Different shapes = different jobs



### **Nucleic acid structure**

#### Monomer= nucleotide

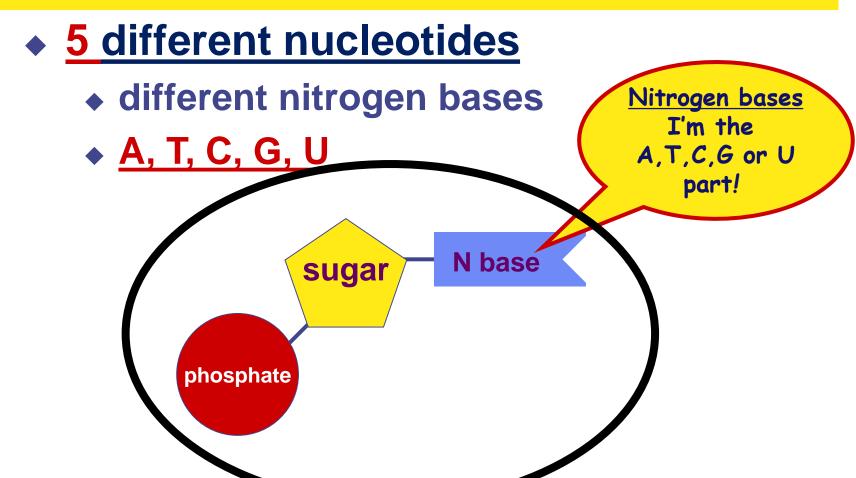
 nucleotides are made of a sugar, phosphate, and nitrogen base



### **Nucleic acids**

■ Monomer = <u>nucleotides</u>

nucleotide – nucleotide – nucleotide – nucleotide



### **Nucleotide chains**

- Nucleic acids
  - nucleotides chained into a polymer
    - DNA
    - RNA

